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Chapter 7, Section 7.2 1. How many  
valence electrons does a neutral  
magnesium (Mg) atom have? 2. What is  
the charge on a magnesium ion? What  
does magnesium have to do to form such  
an ion, and why does it tend to do so? 3.  
How many valence electrons does a single

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neutral chlorine (Cl atom) have? 4.

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Chemistry: Matter and Change Chapter 5

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Energy Levels and Sublevels ...

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Considering your answer to question 5,  
how many electrons can fit in the entire 5s  
energy level? can (A VeI is a ha AC5a)  
5s  $\uparrow$  7. Write the notation for an  
electron spinning clockwise in a p sublevel

...

Grosse Pointe Public School System /

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WORKSHEET The Activity Series Use  
with Chapter 9, Section 9.2 1. For each of  
the following pairs of elements, underline  
the one that would replace the other  
element in a compound. a. calcium, tin e.  
iron, copper b. bromine, fluorine f. iodine,

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chlorine c. aluminum, potassium g. silver,  
lead d. zinc, sodium 2.

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WORKSHEET The Activity Series  
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8 . 1.7, the CF bond is polar covalent. (b)

!ENCI"P = ENCI " ENP. = 3.0" 2.1

!ENCI"P = 0.9. Since  $0.5 < \text{EN. A.} < 1.7$ ,

the PCI bond is polar covalent. (c)

!ENCI"Li Section 8.5 Electronegativity

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polar . that is in excess.

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