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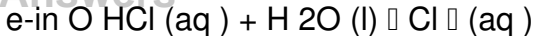
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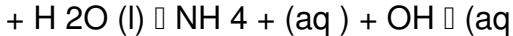
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base, accepting  $H^+$  from  $HCl$  due to attraction between  $H^+$  and lone pair of



+  $H_3O^+(aq)$  Ex. 2: Water acts as

acid, donating  $H^+$  to  $NH_3$   $NH_3(aq)$



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mixture of two or more substances made of a solute, which dissolves in the solvent. A liquid, on the

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