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Chemistry

Chapter 9

Review

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you take that you
require to get those all
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having significantly

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Chapter 3
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CHAPTER 9

HOMEWORK 9-2 (pp.
280 – 282)

VOCABULARY

Complete each sentence. 1. ... Circle the letter of the best answer.

1. Given the balanced equation $2C_4H_{10} + 13O_2 \rightarrow 8CO_2 + 10H_2O$, how many moles of CO_2 are produced when 14.9 g of O_2 are

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used? a. 0.29 mol b.
3.49 mol

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1 Answers

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Stoichiometry Chapter

9 – Stoichiometry

Review #1 – #18,

#31, & #38 Answers .

38. To ensure that all magnesium is converted to MgO , I would use pure oxygen, not air, to carry out the reaction, because Mg could react with N_2 in air to form Mg_3N_2 . The pure oxygen should be in excess. 5. a. 50 mol H_2

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Chapter 9
Page 6/24

Review

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Stoichiometry Section 2

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OpenStax 1. The cutting edge of a knife that has been sharpened has a smaller surface area than a dull knife. Since pressure is force per unit area, a sharp knife will exert a higher pressure with the same amount of force and cut through material more effectively.

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format typically consists
of conceptual questions
and problems to be
worked and answered
(multiple choice format).

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Review Answers H2 + S

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9 Chemical Reactions

Chapter Review F f.

Pentane burns in

oxygen to produce

carbon dioxide and

water $C_5H_{12} + 8O_2$

$\rightarrow 5CO_2 + 6H_2O$

Classify the following

reactions as single

displacement, double

displacement, Page

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